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V. SYNTHESIS, ANTIBACTERIAL ACTIVITY AND ORAL ABSORPTION OF NEW 3-[(Z)-2-METHOXYCARBONYLVINYLTHIO]-7β-[(Z)-2-(2-AMINOTHIAZOL-4-YL)-2-(OXYIMINO)ACETAMIDO]CEPHALOSPORINS

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A series of new 3-[(Z)-2-methoxycarbonylvinylthio]- 7β -[(2-aminothiazol-4-yl)acetamido]cephalosporins (1) having various oxyimino groups (Z-form) at the α position of the C-7 side chain was synthesized and evaluated for antibacterial activity and oral absorption in rats. Of these, the cephalosporin (1a) with a hydroxyimino group in the C-7 side chain showed a potent antibacterial activity against Gram-negative bacteria and Gram-positive *Staphylococcus aureus* as well as good oral absorption in rats. The structure-activity relationships of 1 are also presented.

Recently, we have studied the synthesis and biological properties of cephalosporins with a hetero-atom attached directly to the C-3 position of the cephem nucleus^{1~4)}. In a previous paper⁴⁾, we reported that 7β -[(Z)-2-(2-aminothiazol-4-yl)-2-(carboxymethoxyimino)acetamido]cephalosporins with a lower alkoxy-carbonylvinylthio group (Z-form) at the C-3 position, represented by **2** as shown in Fig. 1, display good oral absorption in rats, potent antibacterial activity against Gram-negative bacteria and improved activity against *Staphylococcus aureus* as compared with cefixime⁵⁾. Subsequently, we studied the chemical modification of the C-7 side chain of **2** in order to find cephalosporins showing both higher activity, especially against *S. aureus* and better oral absorption than **2**.

Herein we describe the synthesis, antibacterial activity and oral absorption in rats of new 3-[(Z)-2-methoxycarbonylvinylthio]cephalosporins (1) having various oxyimino groups (Z-form) at the α position of the 7 β -(2-aminothiazol-4-yl)acetamido side chain.

Chemistry

The new cephalosporins $(1a \sim 11)$ listed in Table 1 were prepared by the synthetic route as shown in Scheme 1. 4-(2-Aminothiazole)acetic acid derivatives (3) having various oxyimino groups at the α position were converted into the acid chlorides with phosphorus pentachloride in dichloromethane at low

Fig. 1. Structure of 1, 2 and cefixime.







temperature. Then, the acid chlorides (not isolated) were reacted with 7β -amino-3-[(Z)-2-methoxycarbonylvinylthio]cephalosporanic acid ester derivatives (4)⁶⁾ in the presence of pyridine to afford the 7β -acylamino derivatives (5). Subsequently, the protecting groups in 5 (except 5i) were removed by a conventional manner (Method A) with trifluoroacetic acid (TFA) and anisole to yield the desired cephalosporins (1). In the case of 5i bearing *O*,*O*-diethylphosphonomethoxyimino group in the C-7 side chain, trimethylsilyl halide was used to hydrolyze the phosphoric acid ester group after treating with TFA-anisole⁷ (Method B).

Antibacterial Activity and Oral Absorption

The *in vitro* antibacterial activities of the new cephalosporins $(1a \sim 1l)$ against selected Gram-positive and Gram-negative bacteria and their peak serum levels after oral administration (50 mg/kg) to rats are summarized in Table 1. For comparison, the MIC values and peak serum levels of 2^{4} and cefixime⁵ are listed at the bottom of Table 1.

Against S. aureus 209P JC-1, most of these new cephalosporins showed improved activity compared with 2, though the analogues 1g and 1i having carboxylic and phosphoric acid as an acidic group,

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Compound			MIC (µg/ml, 10 ⁶ cfu/ml)ª					Peak serum level (µg/ml) ^b	
No.	R ₁	х	S.a.	E.c.	<i>K.p.</i>	M.m.	<i>S.m</i> .	<i>P.m</i> .	po, 50 mg/kg rats ($n = 3$)
1a	Н	Na	0.39	≦0.1	≦0.1	≦0.1	1.56	≦0.1	18.5
1b	CH ₃	Na	1.56	≦0.1	≦0.1	≤ 0.1	1.56	≦0.1	7.4
1c	CH ₂ CH ₂ F	Na	0.78	≦0.1	0.2	≦0.1	0.39	≦0.1	4.1
1d	$CH_2CH=CH_2$	Na	0.39	0.2	0.78	0.2	0.78	0.2	6.5
1e	CH ₂ CO ₂ Et	Na	0.78	1.56	≤ 0.1	0.2	1.56	≤ 0.1	< 3.4
1f	CH_2CONH_2	Na	1.56	≤ 0.1	≦0.1	≦0.1	0.39	≦0.1	4.9
1g	C(CH ₃) ₂ CO ₂ Na	Na	25	1.56	0.39	0.2	1.56	0.2	<1.3
1h	CH2-COONa	Na	3.13	1.56	0.39	0.78	3.13	≦0.1	<2.8
1i	CH ₂ PO ₃ Na ₂	Na	50	0.78	<u>≤</u> 0.1	≦0.1	0.78	≦0.1	< 3.1
1j	сн₂{{ ∥ №-N	Na	6.25	0.78	<u>≤</u> 0.1	≦0.1	1.56	<u>≦</u> 0.1	<2.3
1k		Na	0.39	≦0.1	0.2	0.39	0.39	≦0.1	<2.6
11	$CH_2CH_2NH_2$ (TFA salt)	Н	6.25	0.2	≦0.1	≦0.1	0.39	0.39	3.7
2° Cefixime ^c	CH ₂ CO ₂ Na	Na	12.5 25	0.2 0.78	$ \leq 0.1 \\ \leq 0.1 $	≦0.1 0.2	1.56 0.78	≦0.1 ≦0.1	38.9 28.6

Table 1. In vitro antibacterial activity and peak serum level of $1a \sim 11$.

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0%

The MICs were determined by a standard agar dilution method using Sensitive Test agar (Eiken, Japan).

^b The peak serum levels were measured by a disc-plate method using *Escherichia coli* SC 507 or *Micrococcus luteus* NIHJ as the test organism.

^c For compounds 2 and cefixime see refs 4 and 5, respectively.

Abbreviations: S.a.; Staphylococcus aureus 209P JC-1, E.c.; Escherichia coli NIHJ JC-2, K.p.; Klebsiella pneumoniae IFO 3317, M.m.; Morganella morganii IID 602, S.m.; Serratia marcescens IID 618, P.m.; Proteus mirabilis IFO 3849.

respectively, in the oxime moiety of C-7 side chain exhibited no significant activity. In particular, compounds **1a**, **1d** and **1k** with hydroxyimino, allyloxyimino and 2-aminothiazol-4-ylmethoxyimino group, respectively, in the C-7 side chain showed fairly potent activity.

On the other hand, against the Gram-negative bacteria, these new cephalosporins exhibited a potent antibacterial activity comparable to cefixime and 2, though 1g and 1h were somewhat less active than the others.

According to these results, we found that introduction of the acidic groups (except hydroxyimino group) into the oxime moiety was liable to reduce the activity against *S. aureus*, due to the high hydrophilicity.

In the oral absorption study in rats, only compound 1a exhibited prominent concentrations in serum. However, the peak serum level of 1a was inferior to those of 2 and cefixime. Contrary to our expectation, the oral absorption of 1i, a phosphoric acid analogue of 2, was much lower than that of 2.

The most favorable compound **1a** in this series was then advanced to a preliminary *in vivo* efficacy trial by oral administration. As shown in Table 2, compound **1a** exhibited good efficacy against systemic infections in mice induced by *Klebsiella pneumoniae* 6 and *Escherichia coli* TM-36, but its efficacy was

Organisms	Challenge dose ^a (cfu/mouse)	Compound	ED ₅₀ ^b (mg/kg)	MIC° (µg/ml)
Klebsiella pneumoniae 6	1.0×10^{7}	1a	$3.64 (2.07 \sim 6.59)$	0.025
		2	$0.49 (0.17 \sim 1.48)$	0.05
Escherichia coli TM-36	1.3×10^{7}	1a	6.25 (3.48~11.36)	0.1
		2	1.73 (1.00~ 3.30)	0.2

Table 2. In vivo antibacterial activity of 1a against systemic infections in mice.

Drugs were administered orally 1 hour after infection. Mouse: Male ICR strain, 4 weeks, 10 mice/group.

^a ip, (5% mucin).

^b Probit method (95% confidence limits).

Inoculum size: 10⁶ cfu/ml.

inferior to that of 2, probably due to the lower bioavailability of 1a.

Further biological evaluation of the promising compound **1a**, as well as its prodrugs designed for improving the bioavailability are now under study.

Experimental

IR spectra were taken on a Jasco DS-701G IR spectrometer. ¹H NMR spectra were recorded on a Varian XL-200 NMR spectrometer using TMS or sodium trimethylsilyl propionate- d_4 (in D₂O) as an internal standard. Mass spectra (MS) were measured on a Jeol JMX-DX303 or JMS-SX102 mass spectrometer. Chromatographic separations were done by using Wako Silica gel C-200 (100~200 mesh, Wako, Japan) or Sephadex LH-20 (Pharmacia, Sweden). Analytical HPLC was performed on a TSK gel LS-410 column ($5 \mu m$, $150 \times 4.6 \text{ mm}$, i.d., Tosoh, Japan) eluted with 35% aq acetonitrile containing tetra-*n*-amylammonium bromide (10 mmol) and ammonium acetate (10 mmol), flow rate 1.0 ml/minute at ambient temperature monitoring UV absorbance at 290 nm.

In Vitro and In Vivo Antibacterial Activities

MICs were determined by the 2-fold agar dilution method using Sensitive Test agar (Eiken, Japan) after incubation at 37°C for 18 hours with an inoculum size 10^6 cfu/ml. Mouse protecting experiments were conducted by use of male ICR mice (n=10) infected intraperitoneally with 0.5 ml of a bacterial suspension containing 100% or more minimal lethal doses. Hog gastric mucin (5%, w/v) was added to the suspension before injection. The test drugs in 5% gum arabic were administered orally 1 hour after the infection. Mortality of the animals was recorded daily over a period of 7 days and the ED₅₀ values were calculated by the method of probit⁸).

Oral Absorption Study

Male SLC/Wister rats (n=3) weighing $180 \sim 220$ g were fasted overnight and orally dosed with 50 mg/kg of the test compounds in 5% gum arabic. Serum samples were collected at 0.5, 1, 2, and 4 hours, respectively, after dosing. Serum levels of the test compounds were measured by the disc-plate method using *Escherichia coli* SC 507 or *Micrococcus luteus* NIHJ as a test organism and Sensitive Test agar as the test medium.

Diphenylmethyl 7β -[2-(2-Tritylaminothiazol-4-yl)-2-[(Z)-methoxyimino]acetamido]-3-[(Z)-2-methoxycarbonylvinylthio]-3-cephem-4-carboxylate (5b)

To a solution of 2-(2-tritylaminothiazol-4-yl)-2-[(Z)-methoxyimino]acetic acid (3, $R_2 = Me$, 591 mg, 1.33 mmol) in dry CH₂Cl₂ (16 ml) were successively added pyridine (532 mg, 5.0 equiv) and phosphorus pentachloride (277 mg, 1.0 equiv) at 0°C with stirring, and the reaction mixture was stirred for 20 minutes. Then, a solution of diphenylmethyl 7 β -amino-3-[(Z)-2-methoxycarbonylvinylthio]cephalosporanate (4⁶), $R_3 = CHPh_2$, 560 mg, 0.87 equiv) in dry CH₂Cl₂ (5 ml) was added to the reaction mixture at -10° C and stirred for 30 minutes at 0°C. After the reaction, 0.5% HCl (50 ml) was added to the reaction mixture and extracted with EtOAc (100 ml). The extract was washed with brine (50 ml), dried (MgSO₄) and evaporated.





5		
2		

	Compound		- ¹ H NMR δ (CDCl ₃)		IR (KBr) am^{-1}
No.	R ₂	R_3		(m/z)	cin -
5a	DMB	РМВ	3.12 (1H, d, $J = 17$ Hz), 3.60 (1H, d, $J = 17$ Hz), 3.70 (3H, s), 3.75 (3H, s), 3.77 (3H, s), 3.80 (3H, s), 4.94 (1H, d, $J = 5$ Hz), 5.22 (2H, s), 5.23 (1H, d, $J = 10$ Hz), 5.34 (1H, d, $J = 10$ Hz), 5.81 (1H, dd, $J = 5$, 9 Hz), 5.94 (1H, d, $J = 10$ Hz), 6.35 ~ 6.47 (2H, m), 6.83 (1H, s), 6.86 (2H, d, $J = 8$ Hz), 6.93 (1H, d, J = 10 Hz), 6.98 (1H, br s), 7.20 ~ 7.40 (19H, m)	998 ^b	1790, 1700
5c	CH ₂ CH ₂ F	РМВ	3.49 (1H, d, $J = 18$ Hz), 3.79 (3H, s), 3.80 (1H, d, $J = 18$ Hz), 3.81 (3H, s), 4.52 (2H, dt, $J = 30$, 4Hz), 4.70 (2H, dt, $J = 48$, 4 Hz), 5.09 (1H, d, $J = 5$ Hz), 5.23 (2H, s), 5.92 (1H, d, J = 10 Hz), 5.95 (1H, dd, $J = 5$, 9 Hz), 6.76 (1H, s), 6.82 (1H, d, $J = 9$ Hz), 6.87 (2H, d, $J = 9$ Hz), 6.96 (1H, d, $J = 10$ Hz), 7.00 (1H, s), 7.26~7.40 (17H, m)	893ª	1780, 1680
5d	CH ₂ CH=CH ₂	PMB	3.47 (1H, d, $J = 18$ Hz), 3.77 (3H, s), 3.79 (1H, d, $J = 18$ Hz), 3.80 (3H, s), 4.79 (2H, d, $J = 5$ Hz), 5.09 (1H, d, $J = 5$ Hz), 5.22 (2H, s), 5.25 (1H, d, $J = 12$ Hz), 5.32 (1H, d, $J = 20$ Hz), 5.92 (1H, d, $J = 10$ Hz), 5.92 ~ 6.12 (2H, m), 6.72 (1H, s), 6.87 (2H, d, $J = 9$ Hz), 6.96 (1H, d, $J = 10$ Hz), 7.00 (1H, s), 7.24 ~ 7.42 (17H, m)	887ª	1780, 1680
5e	CH ₂ CO ₂ Et	РМВ	1.26 (3H, t, $J=7$ Hz), 3.49 (1H, d, $J=18$ Hz), 3.88 (3H, s), 3.89 (1H, d, $J=18$ Hz), 3.92 (3H, s), 4.22 (2H, q, $J=7$ Hz), 4.88 (2H, s), 5.09 (1H, d, $J=5$ Hz), 5.24 (2H, s), 5.91 (1H, dd, $J=5$, 9 Hz), 5.92 (1H, d, $J=10$ Hz), 6.82 (1H, s), 6.88 (3H, d, $J=9$ Hz), 6.97 (1H, d, $J=10$ Hz), 7.00 (1H, s), 7.26 ~ 7.40 (17H, m)	933ª	1780, 1725, 1690
5f	CH ₂ CONH ₂	РМВ	3.49 (1H, d, $J = 18$ Hz), 3.76 (1H, d, $J = 18$ Hz), 3.77 (3H, s), 3.80 (3H, s), 4.70 (2H, br s), 5.09 (1H, d, $J = 5$ Hz), 5.21 (2H, s), 5.89 (1H, dd, $J = 5$, 9 Hz), 5.93 (1H, d, $J = 10$ Hz), 6.65 (1H, s), 6.89 (2H, d, $J = 9$ Hz), 6.92 (1H, d, $J = 10$ Hz), 7.00 (1H, br s), 7.20 ~ 7.46 (17H, m), 8.20 (1H, br s)	905 ^b	1780, 1670
5g	C(CH ₃) ₂ CO ₂ Bh	Bh	1.69 (6H, s), 3.27 (1H, d, $J = 17$ Hz), 3.66 (1H, d, $J = 17$ Hz), 3.77 (3H, s), 5.03 (1H, d, $J = 5$ Hz), 5.82 (1H, d, $J = 10$ Hz), 6.00 (1H, dd, $J = 5$, 8 Hz), 6.65 (1H, s), 6.83 (1H, d, $J = 10$ Hz), 6.88 (1H, s), 6.92 (1H, br s), 7.03 (1H, s), 7.14 ~ 7.48 (36H, m)	1,146 ^b	1790, 1740, 1700
5h	CH2-COOBh	РМВ	3.33 (1H, d, $J = 18$ Hz), 3.68 (1H, d, $J = 18$ Hz), 3.78 (3H, s), 3.80 (3H, s), 5.04 (1H, d, $J = 5$ Hz), 5.23 (2H, s), 5.39 (2H, s), 5.90 (1H, d, $J = 10$ Hz), 5.93 (1H, dd, $J = 5$, 9 Hz), 6.72 (1H, d, $J = 9$ Hz), 6.73 (1H, s), 6.87 (2H, d, $J = 9$ Hz), 6.91 (1H, d, $J = 10$ Hz), 7.01 (1H, s), 7.12 (1H, s), 7.26~7.52 (29H, m), 8.13 (2H, d, $J = 8$ Hz)	1,147ª	1780, 1710
5i	CH ₂ PO(OEt) ₂	PMB	1.29 (6H, t, $J = 7$ Hz), 3.48 (1H, d, $J = 18$ Hz), 3.75 (1H, d, $J = 18$ Hz), 3.76 (3H, s), 3.80 (3H, s), 4.05 ~ 4.20 (4H, m), 4.59 (1H, dd, $J = 5$, 18 Hz), 4.68 (1H, dd, $J = 5$, 18 Hz), 5.07 (1H, d, $J = 5$ Hz), 5.21 (2H, s), 5.82 (1H, dd, $J = 5$, 9 Hz), 5.90 (1H, d, $J = 10$ Hz), 6.76 (1H, s), 6.85 (2H, d, $J = 9$ Hz), 6.94 (1H, s), 6.95 (1H, d, $J = 10$ Hz), 7.20 ~ 7.45 (17H, m), 8.62 (1H, d, $J = 9$ Hz)	998 ^ь	1780, 1680

Compound			¹ H NMR δ (CDCL)		IR (KBr)
No.	R ₂	R ₃		(<i>m</i> / <i>z</i>)	cm ⁻¹
5j	СН2Қ №-N DMB	PMB	3.40 (1H, d, $J=17$ Hz), 3.67 (1H, d, $J=17$ Hz), 3.76 (6H, s), 3.79 (3H, s), 3.80 (3H, s), 5.02 (1H, d, $J=5$ Hz), 5.16 (1H, d, $J=12$ Hz), 5.22 (1H, d, $J=12$ Hz), 5.43 (2H, s), 5.45 (1H, d, $J=14$ Hz), 5.52 (1H, d, $J=14$ Hz), 5.85 (1H, dd, $J=5$, 8 Hz), 5.91 (1H, d, $J=10$ Hz), 6.40 ~ 6.50 (2H, m), 6.77 (1H, s), 6.85 (2H, d, $J=8$ Hz), 6.96 (1H, br s), 6.97 (1H, d, J=10 Hz), 7.20 ~ 7.37 (18H, m), 8.56 (1H, d, $J=8$ Hz)	1,080 ^ь	1790, 1700
5k	CH2 NS NH-Tr	РМВ	3.20 (1H, d, $J = 18$ Hz), 3.68 (1H, d, $J = 18$ Hz), 3.78 (3H, s), 3.79 (3H, s), 5.01 and 5.14 (2H, ABq, $J = 12$ Hz), 5.08 (1H, d, $J = 5$ Hz), 5.12 and 5.22 (2H, ABq, $J = 14$ Hz), 5.80 (1H, dd, $J = 5$, 7 Hz), 5.91 (1H, d, $J = 10$ Hz), 6.16 (1H, s), 6.82 (1H, s), 6.83 (2H, d, $J = 9$ Hz), 6.94 (1H, d, $J = 10$ Hz), 7 02 (1H, s), 7.12 ~ 7.40 (32H, m), 9.98 (1H, d, $J = 7$ Hz)	1,202 ^ь	1780, 1690
51	CH ₂ CH ₂ NHBoc	РМВ	1.30 (9H, s), 3.35 \sim 3.50 (2H, m), 3.51 (1H, d, $J=18$ Hz), 3.77 (3H, s), 3.78 (1H, d, $J=18$ Hz), 3.80 (3H, s), 4.36 (2H, t, $J=5$ Hz), 4.98 (1H, t, $J=7$ Hz), 5.08 (1H, d, $J=5$ Hz), 5.22 (2H, s), 5.90 (1H, d, $J=10$ Hz), 5.96 (1H, dd, $J=5$, 9 Hz), 6.64 (1H, s), 6.87 (2H, d, $J=9$ Hz), 6.98 (1H, d, $J=10$ Hz), 7.01 (1H, s), 7.14 \sim 7.40 (17H, m), 8.09 (1H, d, J=9 Hz)	991 ^ь	1780, 1690

Table 3. (Continued)

^a FD, M⁺.

^b FAB or SI-MS, $(M+H)^+$.

Abbreviations: DMB; 2,4-dimethoxybenzyl, PMB; p-methoxybenzyl, Bh; diphenylmethyl, Tr; trityl, Boc; tert-butoxycarbonyl.

The residue was purified by column chromatography on silica gel (eluent; EtOAc - *n*-hexane, 2:3) to yield 243 mg (23%) of **5b** as a pale yellow powder: IR (KBr) cm⁻¹ 1780 (β -lactam), 1720, 1680; ¹H NMR (CDCl₃) δ 3.48 (1H, d, J = 18 Hz, 2-H α), 3.76 (3H, s, CO₂CH₃), 3.81 (1H, d, J = 18 Hz, 2-H β), 4.09 (3H, s, =NOCH₃), 5.12 (1H, d, J = 5 Hz, 6-H), 5.83 (1H, d, J = 10 Hz, =CHCO₂CH₃), 5.96 (1H, dd, J = 5 and 9 Hz, 7-H), 6.76 (1H, s, thiazole 5-H), 6.87 (1H, d, J = 9 Hz, CONH), 6.89 (1H, d, J = 10 Hz, SCH=), 7.02 (2H, s, TrN*H* and CHPh₂), 7.10~7.44 (25H, m, aromatic H); FD-MS *m*/*z* 908 (M⁺).

Similarly, compounds 5a and $5c \sim 5l$ were prepared from 4 with various 4-(2-aminothiazole)acetic acid derivatives 3 according to the procedure described for 5b. Their spectral data are summarized in Table 3.

Sodium 7β -[2-(2-Aminothiazol-4-yl)-2-[(Z)-methoxyimino]acetamido]-3-[(Z)-2-methoxycarbonyl-vinylthio]-3-cephem-4-carboxylate (1b) (Method A)

To a mixture of TFA (3.5 ml) and anisole (0.7 ml) was added compound **5b** (237 mg, 0.26 mmol) under ice-cooling, and the reaction mixture was stirred for 40 minutes at the same temperature. Then, the reaction mixture was added dropwise into a mixture of Et₂O and *n*-hexane (1:2, 40 ml). The precipitated TFA salt of the desired product was collected by filtration. Subsequently, the TFA salt (150 mg) was dissolved in water (5 ml) with NaHCO₃ (44 mg, 2.0 equiv) and chromatographed on Sephadex LH-20 column (eluent; H₂O), and then lyophilized to afford 110 mg (81%) of **1b** as a white amorphous solid: IR (KBr) cm⁻¹ 1760 (β -lactam), 1660; ¹H NMR (D₂O) δ 3.57 (1H, d, J=18 Hz, 2-H α), 3.76 (3H, s, CO₂CH₃), 3.97 (1H, d, J=18 Hz, 2-H β), 4.00 (3H, s, =NOCH₃), 5.30 (1H, d, J=5 Hz, 6-H), 5.85 (1H, d, J=5 Hz, 7-H), 6.04 (1H, d, J=10 Hz, =CHCO₂CH₃), 7.02 (1H, s, thiazole 5-H), 7.31 (1H, d, J=10 Hz, SCH=); HPLC analysis 96% purity.

According to Method A, compounds 1a, $1c \sim 1h$ and $1j \sim 1l$ were prepared from the corresponding cephalosporin derivatives 5. In the case of 1l, the crude TFA salt obtained was purified by Sephadex

Table 4. ¹H NMR and IR spectral data of 1.



Compound			1 NMP δ (D O)	
No.	R ₁	X	$-$ H NMK θ (D ₂ O)	cm ⁻¹
1a	Н	Na	3.57 (1H, d, $J=17$ Hz), 3.77 (3H, s), 3.97 (1H, d, $J=17$ Hz), 5.32 (1H, d, $J=5$ Hz), 5.89 (1H, d, $J=5$ Hz), 6.04 (1H, d, $J=10$ Hz), 7.00 (1H, s) 7.32 (1H, d, $J=10$ Hz)	1770
1c	CH ₂ CH ₂ F	Na	3.58 (1H, d, $J = 17$ Hz), 3.77 (3H, s), 3.98 (1H, d, $J = 17$ Hz), 4.50 (2H, dt, $J = 32$, 4Hz), 4.80 (2H, dt, $J = 48$, 4Hz), 5.32 (1H, d, $J = 5$ Hz), 5.88 (1H, d, $J = 5$ Hz), 6.06 (1H, d, $J = 10$ Hz), 7.07 (1H, s), 7.34 (1H, d, $J = 10$ Hz)	1760
1d	CH ₂ CH=CH ₂	Na	3.59 (1H, d, $J = 18$ Hz), 3.79 (3H, s), 4.00 (1H, d, $J = 18$ Hz), 4.77 (2H, d, $J = 5$ Hz), 5.33 (1H, d, $J = 5$ Hz), 5.35 (1H, d, $J = 11$ Hz), 5.41 (1H, d, $J = 20$ Hz), 5.89 (1H, d, $J = 5$ Hz), 6.07 (1H, d, $J = 11$ Hz), 6.10 (1H, m), 7.06 (1H, s), 7.35 (1H, d, $J = 11$ Hz)	1760
1e	CH ₂ CO ₂ Et	Na	1.28 (3H, t, $J=7$ Hz), 3.56 (1H, d, $J=17$ Hz), 3.77 (3H, s), 3.97 (1H, d, $J=17$ Hz), 4.29 (2H, q, $J=7$ Hz), 4.86 (2H, s), 5.31 (1H, d, $J=5$ Hz), 5.89 (1H, d, $J=5$ Hz), 6.05 (1H, d, $J=10$ Hz), 7.10 (1H, s), 7.34 (1H, d, $J=10$ Hz)	1760
1f	CH ₂ CONH ₂	Na	3.58 (1H, d, $J = 17$ Hz), 3.77 (3H, s), 3.98 (1H, d, $J = 17$ Hz), 4.76 (2H, s), 5.33 (1H, d, $J = 5$ Hz), 5.89 (1H, d, $J = 5$ Hz), 6.06 (1H, d, $J = 10$ Hz), 7.13 (1H, s), 7.34 (1H, d, $J = 10$ Hz)	1760
1g	C(CH ₃) ₂ CO ₂ Na	Na	1.50 (3H, s), 1.52 (3H, s), 3.60 (1H, d, $J = 17$ Hz), 3.78 (3H, s), 3.98 (1H, d, $J = 17$ Hz), 5.33 (1H, d, $J = 5$ Hz), 5.89 (1H, d, $J = 5$ Hz), 6.07 (1H, d, $J = 10$ Hz), 7.02 (1H, s), 7.35 (1H, d, $J = 10$ Hz)	1760
1h	CH2-COONa	Na	3.42 (1H, d, $J = 17$ Hz), 3.77 (3H, s), 3.86 (1H, d, $J = 17$ Hz), 5.24 (1H, d, $J = 5$ Hz), 5.33 (2H, s), 5.83 (1H, d, $J = 5$ Hz), 6.07 (1H, d, $J = 10$ Hz), 7.04 (1H, s), 7.29 (1H, d, $J = 10$ Hz), 7.54 (2H, d, $J = 8$ Hz). 7.90 (2H, d, $J = 8$ Hz)	1770
1j		Na	(DMSO- d_6) 3.38 (1H, d, $J=17$ Hz), 3.62 (3H, s), 3.90 (1H, d, $J=17$ Hz), 5.08 (1H, d, $J=5$ Hz), 5.23 (2H, s), 5.63 (1H, dd, $J=5$, 8 Hz), 5.86 (1H, d, $J=10$ Hz), 6.72 (1H, s), 7.29 (2H, br s), 7.46 (1H, d, $J=10$ Hz)	1770
1k	CH2 N S NH2	Na	(111, d, $J = 10$ Hz), 10.17 (111, d, $J = 0$ Hz) 3.50 (1H, d, $J = 17$ Hz), 3.78 (3H, s), 3.94 (1H, d, $J = 17$ Hz), 5.09 (2H, s), 5.25 (1H, d, $J = 5$ Hz), 5.84 (1H, d, $J = 5$ Hz), 6.07 (1H, d, J = 10 Hz), 6.78 (1H, s), 7.05 (1H, s), 7.33 (1H, d, $J = 10$ Hz)	1760
11	CH ₂ CH ₂ NH ₂ (TFA salt)	Н	3.42 (2H, t, $J=5$ Hz), 3.58 (1H, d, $J=17$ Hz), 3.75 (3H, s), 3.98 (1H, d, $J=17$ Hz), 4.53 (2H, t, $J=5$ Hz), 5.32 (1H, d, $J=5$ Hz), 5.87 (1H, d, $J=5$ Hz), 6.03 (1H, d, $J=10$ Hz), 7.18 (1H, s), 7.30 (1H, d, $J=10$ Hz)	1770

^a β -Lactam.

LH-20 column (eluent; H_2O). The purities of these compounds were 94~97% by HPLC analysis, and their spectral data are summarized in Table 4.

Sodium 7β -[2-(2-Aminothiazol-4-yl)-2-[(Z)-phosphonomethoxyimino]acetamido-3-[(Z)-2-methoxy-carbonylvinylthio]-3-cephem-4-carboxylate (1i) (Method B)

Compound **5i** (569 mg, 0.57 mmol) was added to a mixture of TFA (4.0 ml) and anisole (0.8 ml) under ice-cooling, and stirred for 1 hour at the same temperature. Then, the reaction mixture was added dropwise to a mixture of Et₂O and *n*-hexane (1:2, 40 ml), and the precipitate TFA salt of 7β -[2-(2-aminothiazol-4-yl)-2-[(Z)-0,O-diethylphosphonomethoxyimino]acetamido]-3-[(Z)-2-methoxycarbonylvinylthio]-3-

cephem-4-carboxylic acid (370 mg, 87%) was collected by filtration: IR (KBr) cm⁻¹ 1780 (β -lactam), 1670; ¹H NMR (DMSO- d_6) δ 1.25 (6H, t, J=7Hz, PO(OCH₂CH₃)₂), 3.65 (3H, s, CO₂CH₃), 3.68 (1H, d, J=18 Hz, 2-H α), 4.00 ~ 4.17 (5H, m, PO(OCH₂CH₃)₂ and 2-H β), 4.50 (2H, d, J=7Hz, =NOCH₂), 5.23 (1H, d, J=5 Hz, 6-H), 5.81 (1H, dd, J=5 and 8Hz, 7-H), 6.03 (1H, d, J=10 Hz, =CHCO₂CH₃), 6.80 (1H, s, thiazole 5-H), 7.51 (1H, d, J=10 Hz, SCH=), 9.76 (1H, d, J=8 Hz, CONH).

Subsequently, the above TFA salt (150 mg, 0.2 mmol) was dissolved in dry dichloromethane (10 ml), and *N*,*O*-bis-trimethylsilylacetamide (0.25 ml, 5.0 equiv) was added to the solution at room temperature. Then the mixture was stirred for 20 minutes. To the mixture was added trimethylsilylbromide (0.13 ml, 5.0 equiv), and stirred for 6 hours. After the reaction, the solvent was evaporated *in vacuo*, and the resulting residue was dissolved in water (5 ml), and then the aqueous solution was adjusted to pH 8 with NaHCO₃. After filtration over Celite, the filtrate was chromatographed on Sephadex LH-20 column (eluent; H₂O), and lyophilized to afford 70 mg (54%) of **1i** as an amorphous solid: IR (KBr) cm⁻¹ 1760 (*β*-lactam), 1600; ¹H NMR (D₂O) δ 3.58 (1H, d, *J*=18 Hz, 2-H α), 3.76 (3H, s, CO₂CH₃), 3.98 (1H, d, *J*=18 Hz, 2-H β), 4.27 (2H, d, *J*=7 Hz, =NOCH₂), 5.31 (1H, d, *J*=5 Hz, 6-H), 5.85 (1H, d, *J*=5 Hz, 7-H), 6.05 (1H, d, *J*=10 Hz, =CHCO₂CH₃), 7.02 (1H, s, thiazole 5-H), 7.33 (1H, d, *J*=10 Hz, SCH=); HPLC analysis: 95% purity.

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